CHEMISTRY

Chapter 10

One Mark Questions:

- 1. Why is Group I elements known as the most electropositive element?
- 2. What happens when CaCO3 is subjected to heat?
- 3. Why sulphates of Mg and Be soluble in water?
- 4. Why does beryllium show similarities with Al?
- 5. What is the formula of soda ash?
- 6. Why does table salt get wet in rainy season?
- 7. Why are peroxides and superoxides stable in comparison to other oxides?
- 8. Name the anomalous properties of lithium.

Two Mark Questions:

- 1. Give two uses of sodium.
- 2. Name the elements present in Group 2.
- 3. Give two uses of
 - (i) caustic soda
 - (ii) quick lime
- 4. Beryllium is reducing in nature. Why?
- 5. What is the difference between bakingsoda and baking powder?
- 6. Lithium shows similarities with magnesium in its chemical behavior. Explain